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realize that the procedure described above implies using calculus to find the change in entropy. Chapter 6: Entropy and the Laws of Thermodynamics - Laws of Thermodynamics Zeroth Law: Heat flows from hot to cold First Law: Energy and matter are conserved Second Law: Matter tends towards chaos Third Law: Entropy of a pure crystal at 0K is zero . Internal Energy (\hat{U} . E) and Heat Flow ! Refers to all of the energy contained within a chemical system. Thermodynamics ALL 2014 STUDENT - birdvilleschools.net -

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